High temperature conventional superconductivity. Seven-year journey.

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Abstract

There is dramatic progress in conventional superconductivity since hydrogen sulfide with the critical temperature $T_c = 203\ \text{K}$ was discovered under high pressures of about 150 GPa [1]. Many other superconductors were found some of them even at higher temperatures and high pressures: with $T_c = 243\ \text{K}$ in YH$_9$ [2], 250-260 K in LaH$_{10}$ [3,4], and 287 K in carbonaceous hydrogen sulfide [5]. Two main structures of the superconducting hydrides were discovered. Most of the high temperature hydrides have a cage-like structure. In particular, in lanthanum hydride LaH$_{10}$, La atom is located at the center of the cage of hydrogen atoms. The lanthanum atom acts as an electron donor contributing to electron pairing, while the hydrogen atoms form weak covalent bonds with each other within the cage. This and other superhydrides (YH$_9$, CaH$_6$) can be considered as a close realization of superconducting metallic hydrogen [6]. The second structure realized in H$_3$S$^1$ is different: here each hydrogen atom is connected by a strong covalent bond to the two nearby sulfur atoms. The strong bonding provides large electron-phonon coupling and enhanced superconductivity. The covalent metals are perspective ambient pressure superconductors. We will discuss different ways for further increase of $T_c$ at high and ambient pressures.